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Methyl 4-(5-methoxy-1*H*-indol-3-yl)-benzoate

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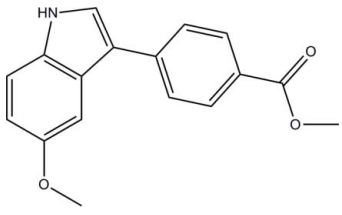
Key indicators: single-crystal X-ray study; T = 113 K; mean σ(C–C) = 0.002 Å; R factor = 0.037; wR factor = 0.110; data-to-parameter ratio = 16.5.

In the title compound, $C_{17}H_{15}NO_3$, the dihedral angle between the benzene ring and the indole ring system is 22.5 (3)°. In the crystal, molecules are linked by N—H···π and C—H···O interactions.

Related literature

For background to the catalysed arylation of indoles, see: Zhang *et al.* (2007). For reference bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$C_{17}H_{15}NO_3$
M_r = 281.30
Monoclinic, $P2_1/c$
a = 15.023 (8) Å
b = 5.871 (3) Å
c = 16.867 (9) Å
β = 113.721 (6)°
V = 1361.9 (12) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.10 mm^{-1}
T = 113 K
0.20 × 0.16 × 0.14 mm

Data collection

Rigaku Saturn724 CCD diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2007)
T_{min} = 0.981, T_{max} = 0.987
13161 measured reflections
3241 independent reflections
2460 reflections with $I > 2\sigma(I)$
R_{int} = 0.038

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.037
$wR(F^2)$ = 0.110
S = 1.06
3241 reflections
196 parameters
1 restraint
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.24 e Å^{-3}
$\Delta\rho_{min}$ = −0.25 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*2 is the centroid ofthe C2–C5/C8/C9 ring.

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···$Cg2^{i}$	0.90 (1)	2.54 (2)	3.295 (2)	142 (1)
C6—H6···$O1^{ii}$	0.95	2.43	3.369 (2)	172
C17—H17B···$O2^{iii}$	0.98	2.60	3.484 (2)	150

Symmetry codes: (i) $-x+1, y-\frac{1}{2}, -z+\frac{3}{2}$; (ii) $x, -y+\frac{3}{2}, z+\frac{1}{2}$; (iii) $x, y+1, z$.

Data collection: *CrystalClear* (Rigaku, 2007); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6526).

References

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supplementary materials

Acta Cryst. (2012). E**68**, o133 [**doi:10.1107/S1600536811052135**]

Methyl 4-(5-methoxy-1*H*-indol-3-yl)benzoate

C.-P. Wang, J.-L. Yu, Z.-Q. Zhang and J.-B. Yan

Comment

In 2007, our group reported direct palladium-catalyzed C-3 arylation of indoles (Zhang *et al.*, 2007). As an extension of this work, we now report the synthesis and crystal structure of the title compound, (I), (Fig. 1).

The dihedral angle between the benzene ring and the indole ring is 22.5 (3)°. All the bond values are within normal ranges (Allen *et al.*, 1987). In the crystal, N—H···π and C—H···O interactions occur (Table 1).

Experimental

A mixture of 5-methoxy-1*H*-indole (0.5 mmol), 1-(4-bromophenyl)ethanone (0.6 mmol), potassium carbonate (1.5 mmol) and $(^{t}Bu)_2P(OH)]_2PdCl_2$ (abbreviated as POPd, 0.025 mmol) was stirred and refluxed in 2 ml of dioxane under nitrogen atmosphere for 24 h. The reaction mixture was allowed to cool to room temperature, quenched with water and extracted with EtOAc. The combined organic layers were washed with brine and dried over $MgSO_4$, and the solvent was removed under vacuum. The residue was purified by chromatography on silica gel eluting with hexane/EtOAc (5/1 by vol.) to give light yellow power in 54.0%, m.p.123.0–124.8 °C. Colourless prisms of (I) were grown by slow evaporation of a solution in chloroform/ethanol (1:1).

Refinement

Atom H1 was located in a difference Fourier map and refined isotropically, with the N—H distance restrained to 0.90 (1) Å. The remaining H atoms were placed in calculated positions (C—H = 0.95–0.98 Å) and refined as riding with U_{iso}(H) = 1.2U_{eq}(C) and 1.5 U_{eq}(C1 and C17).

Figures

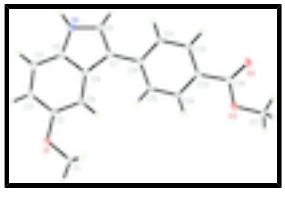


Fig. 1. The molecular structure of the title compound, showing 30% probability displacement ellipsoids for the non-hydrogen atoms.

Methyl 4-(5-methoxy-1*H*-indol-3-yl)benzoate

Crystal data

$C_{17}H_{15}NO_3$

M_r = 281.30

F(000) = 592

D_x = 1.372 Mg m^{-3}

Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 15.023$ (8) Å	Cell parameters from 4649 reflections
$b = 5.871$ (3) Å	$\theta = 1.5$–$27.9°$
$c = 16.867$ (9) Å	$\mu = 0.10$ mm^{-1}
$\beta = 113.721$ (6)°	$T = 113$ K
$V = 1361.9$ (12) Å^3	Prism, colorless
$Z = 4$	0.20 × 0.16 × 0.14 mm

Data collection

Rigaku Saturn724 CCD diffractometer	3241 independent reflections
Radiation source: rotating anode	2460 reflections with $I > 2\sigma(I)$
multilayer	$R_{int} = 0.038$
Detector resolution: 14.22 pixels mm^{-1}	$\theta_{max} = 27.9°$, $\theta_{min} = 1.5°$
ω and φ scans	$h = -19 \rightarrow 19$
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2007)	$k = -7 \rightarrow 7$
$T_{min} = 0.981$, $T_{max} = 0.987$	$l = -22 \rightarrow 22$
13161 measured reflections	

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.037$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.110$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.0702P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
3241 reflections	$(\Delta/\sigma)_{max} = 0.001$
196 parameters	$\Delta\rho_{max} = 0.24$ e Å^{-3}
1 restraint	$\Delta\rho_{min} = -0.25$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.65330 (5)	1.06706 (14)	0.57107 (5)	0.0256 (2)
O2	1.06010 (5)	1.07061 (13)	1.21305 (5)	0.0268 (2)
O3	1.03480 (5)	1.38309 (12)	1.12988 (5)	0.0231 (2)
N1	0.57543 (7)	0.50593 (17)	0.79506 (6)	0.0252 (2)
C1	0.72517 (8)	1.2391 (2)	0.60676 (7)	0.0266 (3)
H1A	0.7069	1.3383	0.6445	0.040*
H1B	0.7300	1.3297	0.5598	0.040*
H1C	0.7881	1.1678	0.6405	0.040*
C2	0.63635 (7)	0.92664 (19)	0.62868 (7)	0.0201 (2)
C3	0.56197 (8)	0.76524 (19)	0.59007 (7)	0.0233 (3)
H3	0.5285	0.7610	0.5288	0.028*
C4	0.53738 (7)	0.61387 (19)	0.64008 (7)	0.0232 (3)
H4	0.4883	0.5024	0.6144	0.028*
C5	0.58711 (7)	0.62954 (18)	0.73024 (7)	0.0206 (2)
C6	0.63929 (8)	0.58338 (19)	0.87366 (7)	0.0230 (2)
H6	0.6447	0.5255	0.9280	0.028*
C7	0.69470 (7)	0.75701 (17)	0.86292 (7)	0.0188 (2)
C8	0.66109 (7)	0.79141 (17)	0.77034 (7)	0.0177 (2)
C9	0.68678 (7)	0.94025 (18)	0.71752 (6)	0.0191 (2)
H9	0.7377	1.0478	0.7424	0.023*
C10	0.77498 (7)	0.86947 (18)	0.93408 (7)	0.0182 (2)
C11	0.82087 (7)	0.75877 (18)	1.01399 (7)	0.0204 (2)
H11	0.7983	0.6129	1.0219	0.024*
C12	0.89807 (7)	0.85678 (18)	1.08147 (7)	0.0202 (2)
H12	0.9277	0.7784	1.1350	0.024*
C13	0.93258 (7)	1.06998 (18)	1.07123 (7)	0.0186 (2)
C14	0.88750 (7)	1.18392 (18)	0.99234 (7)	0.0197 (2)
H14	0.9106	1.3292	0.9845	0.024*
C15	0.80911 (7)	1.08586 (18)	0.92535 (7)	0.0199 (2)
H15	0.7780	1.1670	0.8726	0.024*
C16	1.01559 (7)	1.16901 (18)	1.14560 (7)	0.0194 (2)
C17	1.11194 (8)	1.4968 (2)	1.20010 (7)	0.0237 (3)
H17A	1.1040	1.4704	1.2543	0.036*
H17B	1.1092	1.6607	1.1883	0.036*
H17C	1.1749	1.4365	1.2053	0.036*
H1	0.5318 (9)	0.3932 (19)	0.7867 (10)	0.047 (4)*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0261 (4)	0.0340 (5)	0.0151 (4)	−0.0048 (3)	0.0069 (3)	0.0006 (3)
O2	0.0269 (4)	0.0248 (4)	0.0210 (4)	0.0003 (3)	0.0017 (3)	0.0020 (3)
O3	0.0247 (4)	0.0209 (4)	0.0202 (4)	−0.0035 (3)	0.0056 (3)	−0.0013 (3)
N1	0.0233 (5)	0.0248 (5)	0.0264 (5)	−0.0070 (4)	0.0090 (4)	−0.0006 (4)

C1	0.0273 (6)	0.0322 (7)	0.0208 (6)	−0.0051 (5)	0.0102 (5)	0.0010 (5)
C2	0.0189 (5)	0.0235 (6)	0.0182 (5)	0.0021 (4)	0.0076 (4)	0.0000 (4)
C3	0.0198 (5)	0.0285 (6)	0.0181 (5)	0.0015 (4)	0.0041 (4)	−0.0046 (4)
C4	0.0188 (5)	0.0235 (6)	0.0247 (6)	−0.0020 (4)	0.0058 (4)	−0.0060 (4)
C5	0.0181 (5)	0.0201 (5)	0.0231 (6)	0.0004 (4)	0.0078 (4)	−0.0013 (4)
C6	0.0227 (5)	0.0248 (6)	0.0215 (5)	−0.0009 (4)	0.0090 (4)	0.0018 (4)
C7	0.0188 (5)	0.0190 (5)	0.0190 (5)	0.0011 (4)	0.0081 (4)	−0.0002 (4)
C8	0.0147 (5)	0.0187 (5)	0.0190 (5)	0.0019 (4)	0.0062 (4)	−0.0012 (4)
C9	0.0162 (5)	0.0220 (5)	0.0176 (5)	−0.0008 (4)	0.0052 (4)	−0.0012 (4)
C10	0.0181 (5)	0.0205 (5)	0.0180 (5)	0.0015 (4)	0.0092 (4)	−0.0014 (4)
C11	0.0227 (5)	0.0174 (5)	0.0216 (5)	−0.0001 (4)	0.0095 (4)	0.0007 (4)
C12	0.0222 (5)	0.0201 (6)	0.0179 (5)	0.0041 (4)	0.0075 (4)	0.0023 (4)
C13	0.0194 (5)	0.0202 (6)	0.0175 (5)	0.0025 (4)	0.0087 (4)	−0.0013 (4)
C14	0.0233 (5)	0.0188 (5)	0.0188 (5)	−0.0007 (4)	0.0102 (4)	−0.0006 (4)
C15	0.0239 (5)	0.0208 (6)	0.0158 (5)	0.0010 (4)	0.0089 (4)	0.0015 (4)
C16	0.0196 (5)	0.0198 (5)	0.0204 (5)	0.0018 (4)	0.0097 (4)	−0.0011 (4)
C17	0.0234 (5)	0.0235 (6)	0.0221 (6)	−0.0035 (4)	0.0069 (4)	−0.0031 (4)

Geometric parameters (Å, °)

O1—C2	1.3739 (14)	C6—H6	0.9500
O1—C1	1.4232 (14)	C7—C8	1.4491 (16)
O2—C16	1.2095 (13)	C7—C10	1.4718 (15)
O3—C16	1.3397 (14)	C8—C9	1.4082 (15)
O3—C17	1.4449 (13)	C9—H9	0.9500
N1—C6	1.3643 (15)	C10—C15	1.3997 (16)
N1—C5	1.3811 (15)	C10—C11	1.4025 (15)
N1—H1	0.902 (8)	C11—C12	1.3818 (15)
C1—H1A	0.9800	C11—H11	0.9500
C1—H1B	0.9800	C12—C13	1.3919 (16)
C1—H1C	0.9800	C12—H12	0.9500
C2—C9	1.3828 (15)	C13—C14	1.3970 (15)
C2—C3	1.4089 (16)	C13—C16	1.4859 (15)
C3—C4	1.3742 (16)	C14—C15	1.3872 (15)
C3—H3	0.9500	C14—H14	0.9500
C4—C5	1.4020 (17)	C15—H15	0.9500
C4—H4	0.9500	C17—H17A	0.9800
C5—C8	1.4110 (15)	C17—H17B	0.9800
C6—C7	1.3733 (15)	C17—H17C	0.9800
C2—O1—C1	116.85 (9)	C5—C8—C7	106.77 (9)
C16—O3—C17	115.91 (9)	C2—C9—C8	118.67 (10)
C6—N1—C5	109.30 (10)	C2—C9—H9	120.7
C6—N1—H1	125.4 (10)	C8—C9—H9	120.7
C5—N1—H1	125.3 (10)	C15—C10—C11	117.54 (10)
O1—C1—H1A	109.5	C15—C10—C7	122.41 (10)
O1—C1—H1B	109.5	C11—C10—C7	120.04 (10)
H1A—C1—H1B	109.5	C12—C11—C10	121.59 (10)
O1—C1—H1C	109.5	C12—C11—H11	119.2
H1A—C1—H1C	109.5	C10—C11—H11	119.2

H1B—C1—H1C	109.5	C11—C12—C13	120.22 (10)
O1—C2—C9	123.72 (10)	C11—C12—H12	119.9
O1—C2—C3	114.50 (10)	C13—C12—H12	119.9
C9—C2—C3	121.78 (10)	C12—C13—C14	119.12 (10)
C4—C3—C2	120.68 (10)	C12—C13—C16	118.36 (10)
C4—C3—H3	119.7	C14—C13—C16	122.51 (10)
C2—C3—H3	119.7	C15—C14—C13	120.30 (10)
C3—C4—C5	117.76 (10)	C15—C14—H14	119.8
C3—C4—H4	121.1	C13—C14—H14	119.8
C5—C4—H4	121.1	C14—C15—C10	121.18 (10)
N1—C5—C4	129.97 (11)	C14—C15—H15	119.4
N1—C5—C8	107.51 (10)	C10—C15—H15	119.4
C4—C5—C8	122.51 (10)	O2—C16—O3	123.55 (10)
N1—C6—C7	110.23 (10)	O2—C16—C13	124.46 (10)
N1—C6—H6	124.9	O3—C16—C13	111.99 (9)
C7—C6—H6	124.9	O3—C17—H17A	109.5
C6—C7—C8	106.17 (9)	O3—C17—H17B	109.5
C6—C7—C10	124.52 (10)	H17A—C17—H17B	109.5
C8—C7—C10	129.22 (9)	O3—C17—H17C	109.5
C9—C8—C5	118.56 (10)	H17A—C17—H17C	109.5
C9—C8—C7	134.66 (10)	H17B—C17—H17C	109.5
C1—O1—C2—C9	−1.96 (15)	C5—C8—C9—C2	2.07 (14)
C1—O1—C2—C3	177.66 (9)	C7—C8—C9—C2	−178.95 (11)
O1—C2—C3—C4	179.99 (9)	C6—C7—C10—C15	−159.36 (10)
C9—C2—C3—C4	−0.38 (16)	C8—C7—C10—C15	24.50 (16)
C2—C3—C4—C5	1.45 (16)	C6—C7—C10—C11	21.27 (15)
C6—N1—C5—C4	−179.04 (11)	C8—C7—C10—C11	−154.88 (10)
C6—N1—C5—C8	−0.29 (12)	C15—C10—C11—C12	−1.09 (15)
C3—C4—C5—N1	177.83 (10)	C7—C10—C11—C12	178.32 (9)
C3—C4—C5—C8	−0.75 (16)	C10—C11—C12—C13	−0.27 (15)
C5—N1—C6—C7	−0.45 (13)	C11—C12—C13—C14	0.69 (15)
N1—C6—C7—C8	0.98 (12)	C11—C12—C13—C16	179.85 (9)
N1—C6—C7—C10	−175.91 (9)	C12—C13—C14—C15	0.28 (15)
N1—C5—C8—C9	−179.88 (9)	C16—C13—C14—C15	−178.85 (9)
C4—C5—C8—C9	−1.02 (15)	C13—C14—C15—C10	−1.69 (15)
N1—C5—C8—C7	0.87 (11)	C11—C10—C15—C14	2.06 (15)
C4—C5—C8—C7	179.74 (10)	C7—C10—C15—C14	−177.33 (9)
C6—C7—C8—C9	179.80 (11)	C17—O3—C16—O2	−1.53 (14)
C10—C7—C8—C9	−3.51 (19)	C17—O3—C16—C13	177.50 (8)
C6—C7—C8—C5	−1.13 (11)	C12—C13—C16—O2	5.35 (15)
C10—C7—C8—C5	175.56 (10)	C14—C13—C16—O2	−175.51 (10)
O1—C2—C9—C8	178.17 (9)	C12—C13—C16—O3	−173.67 (8)
C3—C2—C9—C8	−1.43 (15)	C14—C13—C16—O3	5.47 (13)

Hydrogen-bond geometry (Å, °)

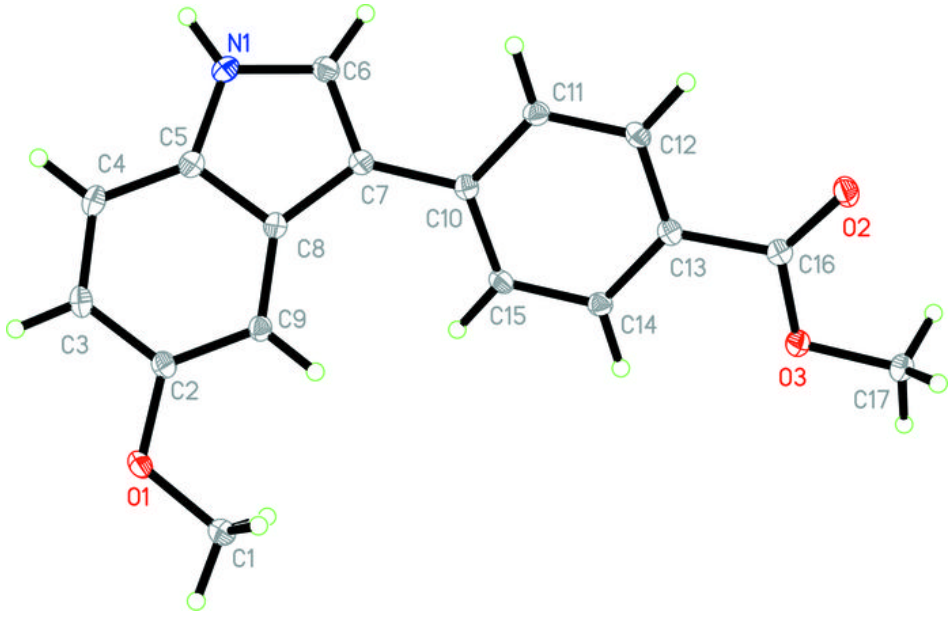
Cg2 is the centroid ofthe C2–C5/C8/C9 ring.

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*

N1—H1···Cg2^{i}	0.90 (1)	2.54 (2)	3.295 (2)	142.(1)
C6—H6···O1^{ii}	0.95	2.43	3.369 (2)	172
C17—H17B···O2^{iii}	0.98	2.60	3.484 (2)	150

Symmetry codes: (i) −*x*+1, *y*−1/2, −*z*+3/2; (ii) *x*, −*y*+3/2, *z*+1/2; (iii) *x*, *y*+1, *z*.

Fig. 1


N1
C6
C11
C5
C12
C4
C7
C10
O2
C8
C13
C16
C15
C14
C9
C3
O3
C17
C2
O1
C1